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# EFFECT OF HOLDING TIME ON THE PRODUCTION OF Nb-NbAl<sub>3</sub> INTERMETALLIC COMPOSITES VIA ELECTRIC-CURRENT-ACTIVATED SINTERING

# VPLIV ČASA ZADRŽANJA NA IZDELAVO Nb-NbAl<sub>3</sub> INTERMETALNIH KOMPOZITOV Z ELEKTRIČNIM TOKOM AKTIVIRANIH S SINTRANJEM

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A recently developed powder metallurgy processing technique – Electric Current Activated (Assisted) Sintering (ECAS) was employed to produce intermetallic Nb-NbAl<sub>3</sub> composites. In this study, to produce Nb-NbAl<sub>3</sub> in-situ intermetallic composites, Nb (99.8 % purity, less than 44  $\mu$ m) and Al (99.5 % purity, less than 44  $\mu$ m) elemental powders were mixed in the stoichiometric ratio corresponding to the Nb-Al phase diagram. The effect of different processing times, for (10, 30, 60) s, under maximum of 2000 A and 1.5-2.0 V, was investigated. Scanning electron microscopy and X-ray diffraction analysis were used to characterize the produced samples. X-ray diffraction studies revealed that the dominant phases are NbAl<sub>3</sub> and Nb. Scanning electron microscopy examinations showed a dense microstructure with a very low amount of porosity and also a trace amount of residual aluminium. The microhardness of the test materials sintered for 60 s via electric-current-activated sintering was about 405 HV±46 HV0,05.

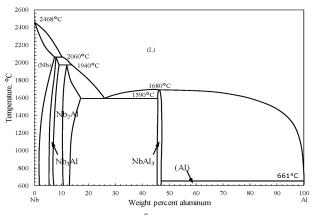
Keywords: in situ composites, NbAl3 aluminides, electric current activated sintering (ECAS)

Pred kratkim razvita tehnika v metalurgiji prahov: sintranje, aktivirano z električnim tokom (ECAS), je bilo uporabljeno za in situ izdelavo intermetalnega kompozita Nb-NbAl<sub>3</sub>. V študiji sta bila za in situ proizvodnjo Nb-NbAl<sub>3</sub> intermetalnega kompozita zmešana elementna prahova Nb (99.8 % čistost, delci manjši od 44 µm) in Al (99.5 % čistost, delci manjši od 44 µm), zmešana v stehiometričnem razmerju, skladno s faznim diagramom Nb-Al. Preiskovan je bil vpliv različnih časov izdelave: (10, 30, 60) s pri toku maksimalno 2000 A in napetosti 1,5-2,0 V. Izdelani vzorci so bili karakterizirani z vrstično elektronsko mikroskopijo in rentgensko difrakcijo. Rentgenska difrakcija je odkrila, da sta prevladujoči fazi NbAl<sub>3</sub> in Nb. Vrstična elektronska mikroskopija je pokazala gosto mikrostrukturo z majhnim deležem poroznosti in sledovi preostalega aluminija. Mikrotrdota preizkusnega materiala, sintranega 60 s, s sintranjem aktiviranim z električnim tokom, je bila okrog 405 HV±46 HV0,05.

Ključne besede: in situ kompoziti, NbAl3 aluminidi, sintranje aktivirano z električnim tokom (ECAS)

#### **1 INTRODUCTION**

Intermetallic compounds have been the focus of significant research and development efforts during recent years. Among intermetallic compounds, niobium aluminides are very important and attractive.<sup>1-3</sup> Three intermetallic compounds are present in the Nb-Al binary system including Nb<sub>3</sub>Al (A15 structure), Nb<sub>2</sub>Al (D8b structure) and NbAl<sub>3</sub> (DO22 structure, TiAl<sub>3</sub> type).<sup>4-7</sup> Among the various compounds in the Nb-Al system (Figure 1), NbAl<sub>3</sub> with its high melting point (1680 °C), low density (4.54 g/cm<sup>3</sup>), is attractive as a potential material for high-temperature applications.<sup>6,8</sup> The applications of NbAl<sub>3</sub> include its use in turbine blades in aircraft engines or in stationary gas turbines.<sup>1,9</sup> However, despite its attractive features, its usage is limited by inadequate ductility at room temperature. So, for optimization of room-temperature toughness, microstructural modifications are required.<sup>1,6,10</sup> In-situ toughening is an alternative technique to enable a combination of brittle intermetallic phase with a ductile metallic phase in one step production.<sup>11,12</sup> Some conventional methods such as melting, casting and mechanical alloying techniques or self-propagating high-temperature synthesis (SHS)<sup>7–9</sup> can be used for manufacturing intermetallics. A recently



**Figure 1:** Nb-Al phase diagram<sup>7</sup> **Slika 1:** Fazni diagram Nb-Al<sup>7</sup>

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developed electric current activated/assisted sintering (ECAS) technique has been used in this study. This system enables the cold formed compact obtained from uniaxial compression to be inserted into a container, which is heated by passing an electrical current. In the present paper we prefer to apply the direct current resistive sintering technique to the Nb+3Al powder mixture in order to investigate a new route to determine the effect of holding time on the production of Nb-NbAl<sub>3</sub> in-situ composites.

# **2 EXPERIMENTAL PART**

# 2.1 Materials and methods

Al and Nb elemental powders with  $35-44 \mu m$  grain size and purity of 99.5 % and 99.8 %, respectively, were mixed to give the nominal composition of Nb40Al60 (*w*/%) for the formation of a NbAl<sub>3</sub> intermetallic based metallic Nb reinforced in-situ composite. The powder mixture was ball milled for 15 min then cold-pressed before sintering to form a cylindrical compact in a metallic die under a uniaxial pressure of 200 MPa. The dimensions of the compact were 15–16 mm diameter and 3–4 mm thickness. The production of Nb aluminidebased intermetallic compound was performed via the electric current activated sintering technique in an open atmosphere at 2000 A for (10, 30, 60) s. They are denoted as C1 and C2 and C3 in the following section, respectively. Th eprocess parameters are listed in **Table 1**.

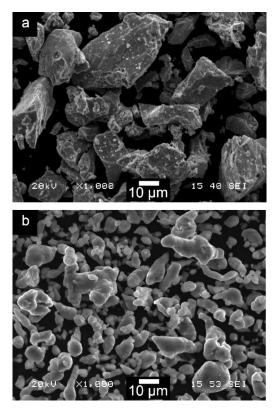


Figure 2: SEM micrographs of: a) Nb, b) Al powder Slika 2: SEM-posnetka: a) Nb in b) Al prahu

Sample code	w/%	Current (A)	Holding time (s)
C1	40Nb-60Al	2000	10
C2	40Nb-60Al	2000	30
C3	40Nb-60Al	2000	60

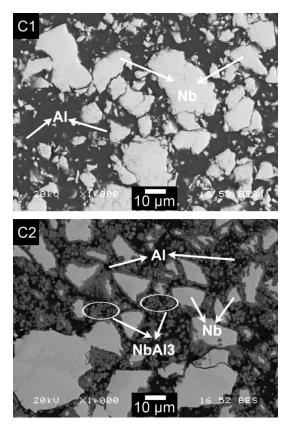
#### 2.2 Characterization

The morphologies of the samples were examined by scanning electron microscopy (SEM-EDS) in terms of the resulting phases. X-ray diffraction (XRD) analyses were carried out using Cu– $K\alpha$  radiation with a wavelength of 0.15418 nm over a  $2\theta$  range of 10–80°. The microhardness of the test materials was measured using a Vickers indentation technique with a load of 0.98 N using Leica WMHT-Mod model Vickers hardness instrument.

# **3 RESULTS AND DISCUSSION**

# 3.1 SEM-EDS Analysis

**Figure 2** shows SEM micrographs of the elemental Nb and Al powder particles. As seen in **Figure 2** the Al particles are rounded and Nb particles are angular and sharp cornered in shape.



**Figure 3:** SEM micrographs of: a) C1 and b) C2 samples **Slika 3:** SEM-posnetka vzorcev a) C1 in b) C2

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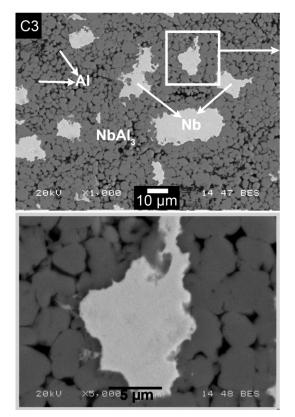


Figure 4: SEM micrographs of C3 Sample: a) 1000 ×, b) 5000 × Slika 4: SEM-posnetka vzorca C3: a) 1000 ×, b) 5000 ×

SEM-EDS analyses of C1, C2 and C3 intermetallic compounds are shown in **Figure 3**. The microstructure in **Figure 3a** shows that the low holding time results in separately formed Nb and Al areas. Increasing the processing time from 10 s to 30 s in C2, (**Figure 3b**), it starts to form a new phase like NbAl<sub>3</sub>, but these microstructures are still far from the desired stoichiometric composition of the main NbAl<sub>3</sub> phase.

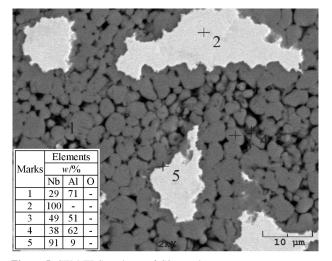


Figure 5: SEM-EDS analyses of C3 samples Slika 5: SEM-EDS-analize na vzorcu C3

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When it comes to the C3 sample, as seen in **Figure 4**, increasing the process holding time to 60 s, the main phase in the microstructure is Nb and NbAl<sub>3</sub>. Besides this, there is also a little amount of residual Al phase and a little oxidation problem because of the open atmosphere sintering in ECAS. But that was not detected in the XRD analyses for being a small amount. In addition to that, a nearly fully dense microstructure was obtained after just a minute of holding process time, thanks to the electric current resistive sintering system.

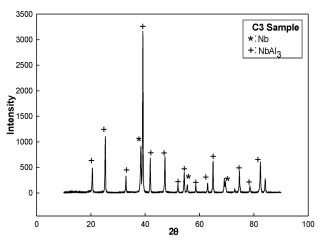
As can be seen in the SEM-EDS analyses in **Figure 5**, the reaction was also not completed in the C3 compound. It is assumed that the applied voltage or current is insufficient for the complete transformation of the NbAl<sub>3</sub> phases in the sintering.

# 3.2 XRD Analysis

The XRD analysis, **Figure 6**, shows that the main phase of the composite is NbAl<sub>3</sub>. The Nb phase is also seen in the XRD analyses, as desired. These results support the observations from the SEM-EDS analysis (**Figure 5**). This can be inferred from this result: Nb-NbAl<sub>3</sub> in-situ composites can be obtained for one minute and one step electric current activated system; however, for eliminating the residual aluminium in the compact. It is obvious that, it can be optimized with some other parameters such as voltage or current.

### 3.3 Hardness

The hardness values  $HV_{0.05}$  of the C1, C2 and C3 samples were measured as  $165\pm20$ ,  $250\pm27$ ,  $405\pm46$  respectively. The hardness of the intermetallic composites increases from  $165\pm20$  HV to  $405\pm46$  HV by increasing the holding time in the process due to the formation of a higher proportion of intermetallic phases. The hardness results for the NbAl<sub>3</sub> composite is in agreement with the literature.<sup>13</sup>



**Figure 6:** XRD Analyses of C3 sample (Nb-NbAl<sub>3</sub>) **Slika 6:** Rentgenogram vzorca C3 (Nb-NbAl<sub>3</sub>)

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#### **4 CONCLUSIONS**

Some of the results obtained from this study can be summarized as follows:

- Niobium aluminide-based composites were fabricated by electric current activated/assisted sintering at 2000 A in only 60 s.
- Under such conditions, the reaction is nearly completed within a very short period of time and the end-product is consolidated to a nearly fully dense microstructure.
- SEM-EDS and XRD analyses showed that 60 s are sufficient for obtaining Nb-NbAl<sub>3</sub> phases. However, for eliminating the residual aluminium in the compact it can be optimized some other parameters such as voltage or current.
- Hardness of the intermetallic composites increases from 165±20 HV to 405±46 HV by increasing the holding time in the process by ensuring the formation of a higher proportion of intermetallic phases

#### Acknowledgement

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#### **5 REFERENCES**

- <sup>1</sup>V. Gauthier, F. Bernard, E. Gaffet, D. Vrel, M. Gailhanou, J. Larpin, Investigations of the formation mechanism of nanostructured NbAl 3 via MASHS reaction, Intermetallics, 10 (**2002**), 377–389, doi:10.1016/S0966-9795(02)00010-9
- <sup>2</sup>L. Murugesh, K. V. Rao, R. Ritchie, Powder processing of ductilephase-toughened Nb- Nb 3 Al in situ composites, Materials Science and Engineering A, 189 (**1994**), 201–208, doi:10.1016/0921-5093(94)90416-2

<sup>3</sup> V. Gauthier, C. Josse, F. Bernard, E. Gaffet, J. Larpin, Synthesis of niobium aluminides using mechanically activated self-propagating high-temperature synthesis and mechanically activated annealing process, Materials Science and Engineering A, 265 (1999), 117–128, doi:10.1016/S0921-5093(98)01141-1

- <sup>4</sup>X. Y. Yan, D. J. Fray, Synthesis of niobium aluminides by electrodeoxidation of oxides, Journal of Alloys and Compounds, 486 (2009), 154–161, doi:10.1016/j.jallcom.2009.06.176
- <sup>5</sup> L. M. Peng, Synthesis and mechanical properties of niobium aluminide-based composites, Materials Science and Engineering A, 480 (2008), 232–236, doi:10.1016/j.msea.2007.07.046
- <sup>6</sup> W. Miao, K. Tao, B. Liu, B. Li, Formation of NbAl 3 by Nb ion implantation using metal vapor vacuum arc ion source, Nuclear Instruments and Methods in Physics Research Section B: Beam Interactions with Materials and Atoms, 160 (2000), 343–348, doi:10.1016/S0168-583X(99)00600-X
- <sup>7</sup> C. L. Yeh, H. J. Wang, Effects of sample stoichiometry of thermitebased SHS reactions on formation of Nb–Al intermetallics, Journal of Alloys and Compounds, 485 (**2009**), 280–284, doi:10.1016/ j.jallcom.2009.06.098
- <sup>8</sup> H. Sina, S. Iyengar, Studies on the formation of aluminides in heated Nb–Al powder mixtures, Journal of Alloys and Compounds, 628 (2015), 9–19, doi:10.1016/j.jallcom.2014.12.151
- <sup>9</sup> N. Wang, C. Du, J. Hou, Y. Zhang, K. Huang, S. Jiao, et al., Direct synthesis of Nb–Al intermetallic nanoparticles by sodiothermic homogeneous reduction in molten salts, Intermetallics, 43 (2013), 45–52, doi:10.1016/j.intermet.2013.07.005
- <sup>10</sup> H. Chung, M. Jilavi, T. Duffey, M. Shannon, W. Kriven, J. Mazumder, NbAl 3/Al microlaminated thin films deposited by UV laser ablation, Thin solid films, 388 (2001), 101–106, doi:10.1016/ S0040-6090(01)00833-1
- <sup>11</sup> T. Yener, S. Okumus, S. Zeytin, In Situ Formation of Ti-TiAl 3 Metallic-Intermetallic Composite by Electric Current Activated Sintering Method, Acta Physica Polonica A, 127 (**2015**), 917–920, doi:10.12693/APhysPolA.127.917
- <sup>12</sup> T. Yener, S. Zeytin, Synthesis And Characterization Of Metallic-Intermetallic Ti-TiAl<sub>3</sub>, Nb-Ti-TiAl<sub>3</sub> Composites Produced With Electric-Current-Activated Sintering (ECAS), Mater. Tehnol., 48 (2014), 847–850
- <sup>13</sup> D. Totten, S. Mackenzie, Handbook of Aluminum vol. 2, Alloy Production and Materials Manufacturing: Markel, Decker, 2003